COMPLETE LISTING OF AMENDED CLAIMS

1. (currently amended) A solid or semisolid preparation which is substantially free of volatile organic solvent, said preparation comprising paroxetine hydrochloride ~~or one of its physiologically acceptable salts~~ in the form of a molecular dispersion in a pharmaceutically acceptable matrix material which comprises a completely synthetic polymer having a glass transition temperature of >90°C in the anhydrous state.

2. (canceled)

3. (currently amended) The preparation of ~~claim 2~~ claim 1 having an active ingredient release of at least 80% after 30 min.

4. (currently amended) A process for producing a solid or semisolid preparation ~~as claimed in claim 1~~ which is substantially free of volatile organic solvent, said preparation comprising paroxetine or one of its physiologically acceptable salts in the form of a molecular dispersion in a pharmaceutically acceptable matrix material which comprises a completely synthetic polymer having a glass transition temperature of >90°C in the anhydrous state,
which process comprises the paroxetine or one of its salts and the matrix material being mixed to give a homogeneous melt in an extruder and subsequently being shaped.

5. (previously presented) The process of claim 4 for producing a paroxetine hydrochloride preparation, wherein paroxetine is processed with ammonium

chloride and the matrix materials to give a homogeneous melt.

6. (previously presented) The process of claim 5, wherein amorphous paroxetine or one of its physiologically acceptable salts is employed.

7. (previously presented) The process of claim 4, wherein the melt is produced at a temperature in the range of 80 to 150°C.

8. (previously presented) The process of claim 4, further comprising applying a vacuum to the extruder while the paroxetine or one of its salts and the matrix material are being mixed if solvents are present therein.

9. (previously presented) The preparation of claim 1, which is also free of water.

10. (previously presented) The preparation of claim 1, wherein the polymer has a glass transition temperature of >90°C to 110°C in the anhydrous state.

11. (previously presented) The preparation of claim 1, wherein the polymer is a copolymer of N-vinylpyrrolidone and vinyl acetate.

12. (previously presented) The preparation of claim 11, wherein the polymer is copovidone.

13. (previously presented) The preparation of claim 1, which is a solid.